## WHAT IS CLAIMED IS:

1	1. A method for transporting elemental sulfur comprising:		
2	(a) combining said elemental sulfur with a non-aqueous liquid carrier		
3	comprising a member selected from the group consisting of anhydrous ammonia and		
4	sulfur dioxide to form a fluid mixture, and		
5	(b) conveying said fluid mixture by way of a transport vessel.		
1	2. The method of claim 1 in which (b) is performed in the absence of any		
2	temperature control of said transport vessel other than by exposure to environmental		
3	conditions.		
1	3. The method of claim 1 in which said fluid mixture is a slurry of solid		
2	elemental sulfur in a liquid solution of sulfur dissolved in said non-aqueous liquid carrier.		
1	4. The method of claim 1 in which said transport vessel is a pipeline, and		
2	(b) comprises pumping said fluid mixture through said pipeline.		
1	5. The method of claim 4 in which said pipeline has an inner surface of		
2	ferrous metal in contact with said fluid mixture.		
1	6. The method of claim 4 in which said pipeline is surrounded by air.		
1	7. The method of claim 4 in which said pipeline is an underground		
2	pipeline.		
1	8. The method of claim 4 in which said pipeline is an underwater		
2	pipeline.		
1	9. The method of claim 1 in which said non-aqueous liquid carrier is		
2	anhydrous ammonia.		
1	10. The method of claim 9 in which said elemental sulfur constitutes at		
2	most about 65% by weight of said fluid mixture.		
1	11. The method of claim 9 in which said elemental sulfur constitutes from		
2	about 20% to about 65% by weight of said fluid mixture.		
1	12. The method of claim 9 in which said elemental sulfur constitutes from		
2	about 40% to about 60% by weight of said fluid mixture.		

1	13.	The method of claim 9 in which said elemental sulfur constitutes from	
2	about 50% to about 6	50% by weight of said fluid mixture.	
1	14.	The method of claim 9 in which (b) is performed at a temperature less	
2	than or equal to 35°C	C.	
1	15.	The method of claim 9 in which (b) is performed at a temperature less	
2	than or equal to 20°C	2.	
1	16.	The method of claim 1 in which said non-aqueous liquid carrier is	
2	sulfur dioxide.		
1	17.	The method of claim 16 in which said elemental sulfur constitutes at	
2	most about 65% by v	veight of said fluid mixture.	
1	18.	The method of claim 16 in which said elemental sulfur constitutes	
2	from about 1,800 pp	m by weight to about 65% by weight of said fluid mixture.	
1	19.	The method of claim 16 in which said elemental sulfur constitutes	
2	from about 1% by w	eight to about 60% by weight of said fluid mixture.	
1	20.	The method of claim 16 in which said elemental sulfur constitutes	
2	from about 10% by v	veight to about 50% by weight of said fluid mixture.	
1	21.	The method of claim 16 in which (b) is performed at a temperature less	
2	than or equal to 40°C	· · · · · · · · · · · · · · · · · · ·	
1	22.	The method of claim 16 in which (b) is performed at a temperature less	
2	than or equal to abou	t 20°C.	
l	23.	A method for extracting elemental sulfur from a sulfur-containing,	
2	substantially water-free geologic formation, said method comprising:		
3		arging said geologic formation with anhydrous ammonia to form a liquid	
+ -		emental sulfur from said geologic formation dissolved in anhydrous	
,	ammonia, and	1	

(b) recovering elemental sulfur from said liquid solution.

1	24. The method of claim 23 further comprising recycling to (a) at least a		
2	portion of said ammonia remaining after recovery of elemental sulfur from said liquid		
3	solution in (b).		
1	25. The method of claim 23 in which said geologic formation is a geologic		
2	mineral formation.		
1	26. A method for extracting elemental sulfur from a sulfur-containing,		
2	substantially water-free mineral formation, said method comprising:		
3	(a) purging said mineral formation with anhydrous ammonia to form a liquid		
4	solution of elemental sulfur from said mineral formation dissolved in anhydrous		
5			
6	(b) recovering elemental sulfur from said liquid solution.		
1	27. The method of claim 26 further comprising recycling to (a) at least a		
2	portion of said ammonia remaining after recovery of elemental sulfur from said liquid		
3	solution in (b).		
_	solution in (c).		
1	28. A method for extracting elemental sulfur from substantially anhydrous		
2	carbonaceous solids, said method comprising:		
3	(a) purging said solids with anhydrous ammonia to form a liquid solution of		
4	elemental sulfur from said solids dissolved in anhydrous ammonia, and		
5	(b) recovering elemental sulfur from said liquid solution.		
1	29. The method of claim 28 further comprising recycling to (a) at least a		
2	portion of the ammonia remaining after recovery of elemental sulfur from said liquid solution		
3	in (b).		
1	30. A method for storing elemental sulfur comprising combining said		
2	elemental sulfur with anhydrous ammonia to form a liquid solution or suspension, and		
3	depositing said solution or suspension in a substantially water-free subterranean formation.		
1	31. The method of claim 30 further comprising withdrawing anhydrous		
2	ammonia from said formation apart from said sulfur, for re-use.		

- 1 32. A composition of matter consisting essentially of a solution or
- 2 suspension formed by combining elemental sulfur with liquid sulfur dioxide.